## A NOVEL TECHNIQUE FOR STUDYING THE EFFECT OF ANTI-BACTERIAL AGENTS ON STAPHYLOCOCCUS AUREUS

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This communication reports the development of a new, physiologically satisfactory method for the simultaneous determination of  $\Delta pH$  &  $\Delta \Psi$ , the osmotic and electrical components respectively of the bacterial proton motive force ( $\Delta p$ ) (Mitchell 1966), and its application in studying the effect of a potent uncoupler, carbonyl cyanide m-chlorophenylhydrazone (CCCP) on  $\Delta p$ .

The addition of glucose to starved, potassium depleted S. aureus (4 x 10 cells ml ) results in the immediate uptake of potassium and the production of acetic acid, a by-product of glucose metabolism. Potassium uptake is believed to occur in response to  $\Delta\Psi$  and by applying a modification of the Nernst equation to the uptake data (Denyer 1979) it was possible to calculate  $\Delta\Psi$ . In addition, since acetic acid will act as a freely permeable acid (Rottenberg 1975), the distribution ratio of the anion across the membrane at equilibrium was used to determine  $\Delta pH$  according to the equation of Rottenberg (1975).

The results obtained can be compared with those determined by other more conventional methods of non-physiological origin (reviewed by Rottenberg 1975).

Table	Comparison	οf	values	for	ДрН	and	ΔΨ	in	<u>s</u> .	aureus	аt	рΗ	7.0.

Component	Technique used for determination								
Component	Acetate Potassium uptake		DMO	DDA <sup>+</sup>	TMPB <sup>+</sup>				
ΔΨ (mv)	-	-148	-	-137	-148				
Δрн	0.82	_	1.00	-	-				

The addition of 50  $\mu M$  CCCP results in the partial collapse of  $\Delta pH$  while a combination of CCCP and valinomycin (1  $\mu g$  ml $^{-1}$ ) produces a marked decrease in both  $\Delta pH$  and  $\Delta \Psi.$ 

The advantage of this new technique lies in its speed, physiological acceptability (acetate is a normal metabolite) and its ability to examine simultaneously changes in both components of  $\Delta p$ . Many preservatives are believed to act in a similar way to CCCP and this technique may provide a useful method for screening for potential preservative action.

DMO - 5,5-Dimethyl-2,4-oxazolidinedione
DDA - Dibenzyldimethylammonium cation
TMPB - Triphenylmethylphosphonium cation
Mitchell, P. (1966) Biol. Rev. 41: 445-502

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